

# Chemistry Chemical Periodicity Study Guide

The Transuranium Elements  
The Periodic Table  
Chemistry of the Elements  
Chemistry, Student Study Guide  
Study Guide for Chemical Principles  
Study Guide to Accompany Russell- General Chemistry  
Grade 9 Chemistry Multiple Choice Questions and Answers (MCQs)  
Introduction to Chemistry  
Survival Guide for General Chemistry with Math Review and Proficiency Questions: How to Get an A  
Chemistry Basics (Speedy Study Guide)  
Study Guide to Accompany Chemistry and Chemical Reactivity  
Study Guide for Hein and Arena's Foundations of College Chemistry  
Chemical Periodicity  
A Level Chemistry Multiple Choice Questions and Answers (MCQs)  
Student's Guide to Fundamentals of Chemistry  
Chemistry & Chemical Reactivity  
Study Guide to Accompany Elements of Chemistry, General, Organic, and Biological -- Robert Boikess/Kenneth Breslauer/Edward Edelson  
Student Study Guide to accompany Foundations of College Chemistry, 11th Edition  
Cracking the AP Chemistry Exam, 2013 Edition  
A-Level Study Guide  
Chemistry Ed H2.2  
Advanced Study Guide  
Chemistry Student Study Guide to Accompany Chemistry  
American Book Publishing Record  
Study Guide to Accompany Chemical Principles, Third Edition  
Study Guide to Accompany General Chemistry  
Student Study Guide to accompany Chemistry  
Study and Problem Solving Guide to Accompany Principles of Modern Chemistry, Oxtoby/Nachtrieb  
The Praxis Series Official Study Guide: Middle School: Science  
Study Guide/Selected Solutions Manual  
CK-12 Chemistry - Second Edition  
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### **The Transuranium Elements**

To accomplish your course goals, use this study guide to enhance your understanding of the text content and to be better prepared for quizzes and tests. This convenient manual helps you assimilate and master the information encountered in the text through the use of practice exercises and applications, comprehensive review tools, and additional helpful resources.

### **The Periodic Table**

### **Chemistry of the Elements**

This is an ebook version of the "A-Level Study Guide - Chemistry (Higher 2) - Ed

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H2.2" published by Step-by-Step International Pte Ltd. [ For the revised Higher 2 (H2) syllabus with first exam in 2017. ] This ebook gives concise illustrated notes and worked examples. It is intended as a study guide for readers who have studied the O-Level Chemistry or the equivalent. It contains material that most readers should want to take note of when attending formal lessons and/or discussions on the Singapore-Cambridge GCE A-Level Higher 2 (H2) Chemistry. [As the Higher 1 (H1) Chemistry syllabus is a subset of the H2 Chemistry syllabus, this ebook is also suitable for readers studying Chemistry at the H1 level.] The concise notes cover essential steps to understand the relevant theories. The illustrations and worked examples show essential workings to apply those theories. We believe the notes and illustrations will help readers learn to "learn" and apply the relevant knowledge. The ebook should help readers study and prepare for their exams. Relevant feedbacks from Examiner Reports, reflecting what the examiners expected, are incorporated into the notes and illustrations where possible, or appended as notes (NB) where appropriate. It is also a suitable aid for teaching and revision.

### **Chemistry, Student Study Guide**

### **Study Guide for Chemical Principles**

## **Study Guide to Accompany Russell- General Chemistry**

### **Grade 9 Chemistry Multiple Choice Questions and Answers (MCQs)**

Work more effectively and gauge your progress along the way! This Student Study Guide that accompanies Hein's Foundations of College Chemistry, 11th Edition, contains a review and self-evaluation section, challenge problems, and answers and solutions for all exercises in the study guide. Foundations of College Chemistry is the book that defined the prep/intro chemistry market 35 years ago! And it's been a market leader ever since! Hein/Arena is known for its accuracy, clear no-nonsense approach, and direct writing style. Strong problem solving and carefully constructed problem sets make this book a standout among its many imitators.

### **Introduction to Chemistry**

### **Survival Guide for General Chemistry with Math Review and Proficiency Questions: How to Get an A**

### **Chemistry Basics (Speedy Study Guide)**

A study guide of chemistry basics would give students bite sized chunks of information on a complex subject, making the task of learning it easier and more enjoyable. Study guides present detailed information on a topic in a compact and simplistic way, which helps the student grasp more of what they are being taught. Also, the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory.

### **Study Guide to Accompany Chemistry and Chemical Reactivity**

### **Study Guide for Hein and Arena's Foundations of College Chemistry**

Provides techniques for achieving high scores on the AP chemistry exam and includes two full-length practice tests, a subject review for all topics, and sample questions and answers.

## **Chemical Periodicity**

### **A Level Chemistry Multiple Choice Questions and Answers (MCQs)**

### **Student's Guide to Fundamentals of Chemistry**

### **Chemistry & Chemical Reactivity**

### **Study Guide to Accompany Elements of Chemistry, General, Organic, and Biological -- Robert Boikess/Kenneth Breslauer/Edward Edelson**

Designed for students in Nebo School District, this text covers the Utah State Core Curriculum for chemistry with few additional topics.

## **Student Study Guide to accompany Foundations of College Chemistry, 11th Edition**

## **Cracking the AP Chemistry Exam, 2013 Edition**

## **A-Level Study Guide Chemistry Ed H2.2**

This third edition continues to innovate by providing students with an integrated and modern approach to the subject. The text emphasizes the modern tools of chemistry while incorporating historical evidence, and its unique molecular/quantitative emphasis is further reinforced by an integrated media package developed by the authors. Also of benefit is the just-in-time presentation of key content - only providing details once they are needed. While key topics and analytical techniques have been updated, there is now an additional, third chapter on chemical equilibrium. The authors have also developed an expanded and more integrated problem-solving emphasis that now incorporates a 4-step strategy throughout, complete with text icons. The whole is backed by a range of supplements, including a new illustration program, a tutorial CD, interactive learningware, an extensive Web CT component, an instructor's resource CD, and a

solution CD.

### **Advanced Study Guide Chemistry**

### **Student Study Guide to Accompany Chemistry**

Reviews essential points from the text, presents problems, and hints to assist the student in solving them. Solutions to selected problems in the end of chapter problem sets are included. By Kenneth Hughes of the University of Wisconsin-Madison.

### **American Book Publishing Record**

### **Study Guide to Accompany Chemical Principles, Third Edition**

CK-12 Foundation's Chemistry - Second Edition FlexBook covers the following chapters: Introduction to Chemistry - scientific method, history. Measurement in Chemistry - measurements, formulas. Matter and Energy - matter, energy. The Atomic Theory - atom models, atomic structure, sub-atomic particles. The Bohr

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Model of the Atom electromagnetic radiation, atomic spectra. The Quantum Mechanical Model of the Atom energy/standing waves, Heisenberg, Schrodinger. The Electron Configuration of Atoms Aufbau principle, electron configurations. Electron Configuration and the Periodic Table- electron configuration, position on periodic table. Chemical Periodicity atomic size, ionization energy, electron affinity. Ionic Bonds and Formulas ionization, ionic bonding, ionic compounds. Covalent Bonds and Formulas nomenclature, electronic/molecular geometries, octet rule, polar molecules. The Mole Concept formula stoichiometry. Chemical Reactions balancing equations, reaction types. Stoichiometry limiting reactant equations, yields, heat of reaction. The Behavior of Gases molecular structure/properties, combined gas law/universal gas law. Condensed Phases: Solids and Liquids intermolecular forces of attraction, phase change, phase diagrams. Solutions and Their Behavior concentration, solubility, colligative properties, dissociation, ions in solution. Chemical Kinetics reaction rates, factors that affect rates. Chemical Equilibrium forward/reverse reaction rates, equilibrium constant, Le Chatelier's principle, solubility product constant. Acids-Bases strong/weak acids and bases, hydrolysis of salts, pH Neutralization dissociation of water, acid-base indicators, acid-base titration, buffers. Thermochemistry bond breaking/formation, heat of reaction/formation, Hess' law, entropy, Gibb's free energy. Electrochemistry oxidation-reduction, electrochemical cells. Nuclear Chemistry radioactivity, nuclear equations, nuclear energy. Organic Chemistry straight chain/aromatic hydrocarbons, functional

groups.Chemistry Glossary

## **Study Guide to Accompany General Chemistry**

### **Student Study Guide to accompany Chemistry**

"A Level Chemistry Multiple Choice Questions and Answers (MCQs): Quizzes & Practice Tests with Answer Key" provides mock tests for competitive exams to solve 1745 MCQs. "A Level Chemistry MCQ" pdf to download helps with theoretical, conceptual, and analytical study for self-assessment, career tests. A level chemistry quizzes, a quick study guide can help to learn and practice questions for placement test preparation. "A Level Chemistry Multiple Choice Questions and Answers" pdf to download is a revision guide with a collection of trivia quiz questions and answers pdf on topics: Alcohols and esters, atomic structure and theory, benzene, chemical compound, carbonyl compounds, carboxylic acids, acyl compounds, chemical bonding, chemistry of life, electrode potential, electrons in atoms, enthalpy change, equilibrium, group IV, groups II and VII, halogenoalkanes, hydrocarbons, introduction to organic chemistry, ionic equilibria, lattice energy, moles and equations, nitrogen and sulfur, organic and nitrogen compounds, periodicity, polymerization, rates of reaction, reaction kinetics, redox reactions and

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electrolysis, states of matter, transition elements to enhance teaching and learning. A Level Chemistry Quiz Questions and Answers pdf also covers the syllabus of many competitive papers for admission exams of different universities from chemistry textbooks on chapters: Alcohols and Esters MCQs: 27 Multiple Choice Questions. Atomic Structure and Theory MCQs: 37 Multiple Choice Questions. Benzene: Chemical Compound MCQs: 41 Multiple Choice Questions. Carbonyl Compounds MCQs: 29 Multiple Choice Questions. Carboxylic Acids and Acyl Compounds MCQs: 27 Multiple Choice Questions. Chemical Bonding MCQs: 213 Multiple Choice Questions. Chemistry of Life MCQs: 29 Multiple Choice Questions. Electrode Potential MCQs: 62 Multiple Choice Questions. Electrons in Atoms MCQs: 53 Multiple Choice Questions. Enthalpy Change MCQs: 45 Multiple Choice Questions. Equilibrium MCQs: 50 Multiple Choice Questions. Group IV MCQs: 53 Multiple Choice Questions. Groups II and VII MCQs: 180 Multiple Choice Questions. Halogenoalkanes MCQs: 33 Multiple Choice Questions. Hydrocarbons MCQs: 53 Multiple Choice Questions. Introduction to Organic Chemistry MCQs: 52 Multiple Choice Questions. Ionic Equilibria MCQs: 56 Multiple Choice Questions. Lattice Energy MCQs: 33 Multiple Choice Questions. Moles and Equations MCQs: 50 Multiple Choice Questions. Nitrogen and Sulfur MCQs: 89 Multiple Choice Questions. Organic and Nitrogen Compounds MCQs: 54 Multiple Choice Questions. Periodicity MCQs: 202 Multiple Choice Questions. Polymerization MCQs: 36 Multiple Choice Questions. Rates of Reaction MCQs: 39 Multiple Choice Questions. Reaction Kinetics MCQs: 52 Multiple Choice Questions. Redox Reactions and Electrolysis

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MCQs: 55 Multiple Choice Questions. States of Matter MCQs: 66 Multiple Choice Questions. Transition Elements MCQs: 29 Multiple Choice Questions. "Alcohols and Esters MCQs" pdf covers quiz questions about introduction to alcohols, and alcohols reactions. "Atomic Structure and Theory MCQs" pdf covers quiz questions about atom facts, elements and atoms, number of nucleons, protons, electrons, and neutrons. "Benzene: Chemical Compound MCQs" pdf covers quiz questions about introduction to benzene, arenes reaction, phenol and properties, and reactions of phenol. "Carbonyl Compounds MCQs" pdf covers quiz questions about introduction to carbonyl compounds, aldehydes and ketone testing, nucleophilic addition with HCN, preparation of aldehydes and ketone, reduction of aldehydes, and ketone. "Carboxylic Acids and Acyl Compounds MCQs" pdf covers quiz questions about acidity of carboxylic acids, acyl chlorides, ethanoic acid, and reactions to form tri-iodomethane. "Chemical Bonding MCQs" pdf covers quiz questions about chemical bonding types, chemical bonding electron pair, bond angle, bond energy, bond energy, bond length, bonding and physical properties, bonding energy, repulsion theory, covalent bonding, covalent bonds, double covalent bonds, triple covalent bonds, electron pair repulsion and bond angles, electron pair repulsion theory, enthalpy change of vaporization, intermolecular forces, ionic bonding, ionic bonds and covalent bonds, ionic bonds, metallic bonding, metallic bonding and delocalized electrons, number of electrons, sigma bonds and pi bonds, sigma-bonds, pi-bonds, s-orbital and p-orbital, Van der Waals forces, and contact points. "Chemistry of Life MCQs" pdf covers quiz questions

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about introduction to chemistry, enzyme specificity, enzymes, reintroducing amino acids, and proteins. "Electrode Potential MCQs" pdf covers quiz questions about electrode potential, cells and batteries, e-plimsoll values, electrolysis process, measuring standard electrode potential, quantitative electrolysis, redox, and oxidation. "Electrons in Atoms MCQs" pdf covers quiz questions about electronic configurations, electronic structure evidence, ionization energy, periodic table, simple electronic structure, sub shells, and atomic orbitals. "Enthalpy Change MCQs" pdf covers quiz questions about standard enthalpy changes, bond energies, enthalpies, Hess law, introduction to energy changes, measuring enthalpy changes. "Equilibrium MCQs" pdf covers quiz questions about equilibrium constant expression, equilibrium position, acid base equilibria, chemical industry equilibria, ethanoic acid, gas reactions equilibria, and reversible reactions. "Group IV MCQs" pdf covers quiz questions about introduction to group IV, metallic character of group IV elements, ceramic, silicon oxide, covalent bonds, properties variation in group IV, relative stability of oxidation states, and tetra chlorides. "Groups II and VII MCQs" pdf covers quiz questions about atomic number of group II metals, covalent bonds, density of group II elements, disproportionation, fluorine, group II elements and reactions, group VII elements and reactions, halogens and compounds, ionic bonds, melting points of group II elements, metallic radii of group II elements, periodic table elements, physical properties of group II elements, physical properties of group VII elements, reaction of group II elements with oxygen, reactions of group II elements, reactions of group VII elements,

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thermal decomposition of carbonates and nitrates, thermal decomposition of group II carbonates, thermal decomposition of group II nitrates, uses of group II elements, uses of group II metals, uses of halogens and their compounds. "Halogenoalkanes MCQs" pdf covers quiz questions about halogenoalkanes, uses of halogenoalkanes, elimination reactions, nucleophilic substitution in halogenoalkanes, and nucleophilic substitution reactions. "Hydrocarbons MCQs" pdf covers quiz questions about introduction to alkanes, sources of alkanes, addition reactions of alkenes, alkanes reaction, alkenes and formulas. "Introduction to Organic Chemistry MCQs" pdf covers quiz questions about organic chemistry, functional groups, mechanisms, organic reactions, naming organic compounds, stereoisomerism, structural isomerism, and types of organic reactions. "Ionic Equilibria MCQs" pdf covers quiz questions about introduction to ionic equilibria, buffer solutions, equilibrium and solubility, indicators and acid base titrations, pH calculations, and weak acids. "Lattice Energy MCQs" pdf covers quiz questions about introduction to lattice energy, ion polarization, lattice energy value, atomization and electron affinity, Born Haber cycle, and enthalpy changes in solution. "Moles and Equations MCQs" pdf covers quiz questions about amount of substance, atoms, molecules mass, chemical formula and equations, gas volumes, mole calculations, relative atomic mass, solutions, and concentrations. "Nitrogen and Sulfur MCQs" pdf covers quiz questions about nitrogen gas, nitrogen and its compounds, nitrogen and gas properties, ammonia, ammonium compounds, environmental problems caused by nitrogen compounds and nitrate fertilizers, sulfur and oxides, sulfuric acid and

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properties, and uses of sulfuric acid. "Organic and Nitrogen Compounds MCQs" pdf covers quiz questions about amides in chemistry, amines, amino acids, peptides and proteins. "Periodicity MCQs" pdf covers quiz questions about acidic oxides, basic oxides, aluminum oxide, balancing equation, period 3 chlorides, balancing equations: reactions with chlorine, balancing equations: reactions with oxygen, bonding nature of period 3 oxides, chemical properties of chlorine, chemical properties of oxygen, chemical properties periodicity, chemistry periodic table, chemistry: oxides, chlorides of period 3 elements, electrical conductivity in period 3 oxides, electronegativity of period 3 oxides, ionic bonds, molecular structures of period 3 oxides, oxidation number of oxides, oxidation numbers, oxides and hydroxides of period 3 elements, oxides of period 3 elements, period III chlorides, periodic table electronegativity, physical properties periodicity, reaction of sodium and magnesium with water, and relative melting point of period 3 oxides.

"Polymerization MCQs" pdf covers quiz questions about types of polymerization, polyamides, polyesters, and polymer deductions. "Rates of Reaction MCQs" pdf covers quiz questions about catalysis, collision theory, effect of concentration, reaction kinetics, and temperature effect on reaction rate. "Reaction Kinetics MCQs" pdf covers quiz questions about reaction kinetics, catalysts, kinetics and reaction mechanism, order of reaction, rate constant  $k$ , and rate of reaction.

"Redox Reactions and Electrolysis MCQs" pdf covers quiz questions about redox reaction, electrolysis technique, oxidation numbers, redox and electron transfer.

"States of Matter MCQs" pdf covers quiz questions about states of matter,

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ceramics, gaseous state, liquid state, materials conservations, solid state.  
"Transition Elements MCQs" pdf covers quiz questions about transition element, ligands and complex formation, physical properties of transition elements, redox and oxidation.

### **Study and Problem Solving Guide to Accompany Principles of Modern Chemistry, Oxtoby/Nachtrieb**

Study Guide/Selected Solutions Manual to accompany Fundamentals of Chemistry contains a brief overview of every chapter, review of skills, self tests and the answers and detailed solutions to all odd-numbered end-of-chapter problems in the text book.

### **The Praxis Series Official Study Guide: Middle School: Science**

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Study more effectively and improve your performance at exam time with this comprehensive guide. The guide includes chapter summaries that highlight the main themes; study goals with section references; lists of important terms; a preliminary test for each chapter that provides an average of 80 drill and concept questions; and answers to the preliminary tests. The Study Guide helps you organize the material and practice applying the concepts of the core text. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

### **Directory of Distance Learning Opportunities**

This survival guide focuses on helping students practice for exams and shows them how to solve difficult problems by dissecting them into manageable chunks. Written in the style of a student meeting with an instructor during office hours, it addresses the most frequently asked questions. This approach leads to the three levels approach - A, B, and minimal - to dissect a typical difficult question into manageable chunks and quickly build student confidence to master the knowledge needed to succeed in the course. This book is available for students to purchase at [www.CENGAGEbrain.com](http://www.CENGAGEbrain.com) or available for packaging with any Cengage textbook. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

### **Study Guide to Accompany Chemistry**

The periodic table of elements, first encountered by many of us at school, provides an arrangement of the chemical elements, ordered by their atomic number, electron configuration, and recurring chemical properties, and divided into periodic trends. In this Very Short Introduction Eric R. Scerri looks at the trends in properties of elements that led to the construction of the table, and shows how the deeper meaning of the table's structure gradually became apparent with the development of atomic theory and, in particular, quantum mechanics, which underlies the behaviour of all of the elements and their compounds. This new edition, publishing in the International Year of the Periodic Table, celebrates the completion of the seventh period of the table, with the ratification and naming of elements 113, 115, 117, and 118 as nihonium, moscovium, tennessine, and oganesson. Eric R. Scerri also incorporates new material on recent advances in our understanding of the origin of the elements, as well as developments concerning group three of the periodic table. ABOUT THE SERIES: The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area. These pocket-sized books are the perfect way to get ahead in a new subject quickly. Our expert authors combine facts, analysis, perspective, new ideas, and enthusiasm to make interesting and challenging topics highly readable.

### **Study Guide and Glossary to Accompany Chemical Principles**

"Grade 9 Chemistry Multiple Choice Questions and Answers (MCQs): Quizzes & Practice Tests with Answer Key" provides mock tests for competitive exams to solve 230 MCQs. "Grade 9 Chemistry MCQ" pdf to download helps with theoretical, conceptual, and analytical study for self-assessment, career tests. Grade 9 chemistry quizzes, a quick study guide can help to learn and practice questions for placement test preparation. "Grade 9 Chemistry Multiple Choice Questions and Answers" pdf to download is a revision guide with a collection of trivia quiz questions and answers pdf on topics: Chemical reactivity, electrochemistry, fundamentals of chemistry, periodic table and periodicity, physical states of matter, solutions, structure of atoms, structure of molecules to enhance teaching and learning. Grade 9 Chemistry Quiz Questions and Answers pdf also covers the syllabus of many competitive papers for admission exams of different schools from chemistry textbooks on chapters: Chemical Reactivity MCQs: 18 Multiple Choice Questions. Electrochemistry MCQs: 32 Multiple Choice Questions. Fundamentals of Chemistry MCQs: 50 Multiple Choice Questions. Periodic Table and Periodicity MCQs: 25 Multiple Choice Questions. Physical States of Matter MCQs: 30 Multiple Choice Questions. Solutions MCQs: 30 Multiple Choice Questions. Structure of Atoms MCQs: 20 Multiple Choice Questions. Structure of Molecules MCQs: 25 Multiple Choice Questions. "Chemical Reactivity MCQs" pdf covers quiz questions about metals, and non-metals. "Electrochemistry MCQs" pdf covers quiz questions

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about corrosion and prevention, electrochemical cells, electrochemical industries, oxidation and reduction, oxidation reduction and reactions, oxidation states, oxidizing and reducing agents. "Fundamentals of Chemistry MCQs" pdf covers quiz questions about atomic and mass number, avogadro number and mole, branches of chemistry, chemical calculations, elements and compounds particles, elements compounds and mixtures, empirical and molecular formulas, gram atomic mass molecular mass and gram formula, ions and free radicals, molecular and formula mass, relative atomic mass, and mass unit. "Periodic Table and Periodicity MCQs" pdf covers quiz questions about periodic table, periodicity and properties. "Physical States of Matter MCQs" pdf covers quiz questions about allotropes, gas laws, liquid state and properties, physical states of matter, solid state and properties, types of bonds, and typical properties. "Solutions MCQs" pdf covers quiz questions about aqueous solution solute and solvent, concentration units, saturated unsaturated supersaturated and dilution of solution, solubility, solutions suspension and colloids, and types of solutions. "Structure of Atoms MCQs" pdf covers quiz questions about atomic structure experiments, electronic configuration, and isotopes. "Structure of Molecules MCQs" pdf covers quiz questions about atoms reaction, bonding nature and properties, chemical bonds, intermolecular forces, and types of bonds.

### **Study Guide for Chang's Chemistry**

## **Student study guide**

## **Manual of Chemistry**

## **Study Guide for Whitten/Davis/Peck/Stanley's Chemistry, 10th**

## **Study Guide for Chemical Principles, Fourth Edition, by Dickerson, Gray, Darensbourg, and Darensbourg**

## **Study Guide for E. Russell Hardwick's Introduction to Chemistry**

Offers information on more than six thousand K-12 courses and programs offered through correspondence or electronic delivery systems in the United States.

## **Approach to Chemistry**

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This is an ebook version of the "Advanced Study Guide - Chemistry - Ed 1.0" published by Step-by-Step International Pte Ltd. [ For the Higher 2 (H2) syllabus with last exam in 2016.] This ebook gives concise illustrated notes and worked examples. It is organised largely accordingly to the Singapore-Cambridge GCE A-Level Higher 2 (H2) syllabus, with additional topics to cover the equivalent syllabuses of the University of Cambridge International Examination (CIE) A Level (Core & A2), and the International Baccalaureate (IB) Higher Level (Core & AHL). The concise notes cover essential steps to understand the relevant theories. The illustrations and worked examples show essential workings to apply those theories. We believe the notes and illustrations will help readers learn to "learn" and apply the relevant knowledge. The ebook should help readers study and prepare for their exams. Relevant feedbacks from Examiner Reports, reflecting what the examiners expected, are incorporated into the notes and illustrations where possible, or appended as notes (NB) where appropriate. It is also a suitable aid for teaching and revision. Sample pages are available (in .pdf) from our website.

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